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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 134 pages. Original publisher: Ada, OK : U. S. Environmental Protection Agency, 1995. LC Number: TD1040 . B46 1995 OCLC Number: (OCoLC)33825557 Subject: Hazardous waste site remediation -- United States. Excerpt: . . . iron fraction in the absence of sulfate reduction. Wet rather than the cis isomer is indicative of an abiotic sediment samples were extracted with 5 M HCl for a dechlorination mechanism (by Fe (II)) remains to be elu-21-day period, and analyzed according to Heron and cidated. VC did not dechlorinate under sulfate-reducing Christensen (5). conditions during the 2 months monitored to date. This information was then used in conjunction with avail- Information available from previous field studies in col-able data on ground-water chemistry (pH, redox, car- laboration with the Robert S. Kerr Laboratories (Dr. John bonate, sulfate, sulfide) and compiled in a chemical Wilson) suggested that neither methanotrophic nor equilibrium model (MINEQL) to predict speciation of methanogenic activity predominates in the contaminant iron. Precipitation of ferrous iron solids was based on plume under the shoreline and Lake Michigan, based on stability constants from the literature. redox potential and oxygen and methane measure-ments. Redox potentials and an increase in soluble iron Results concentrations between both shore samples and below the Lake Michigan bottom, however, suggest that iron The analysis of samples from wells 55AB and 55AD reduction may be the dominant process in regions near indicated the presence of similar concentrations of plume emergence (8). bioavailable Fe (III) (5 mmol equiv. kg) (Table 1). Whereas precipitated iron oxides and ferrous monosul-References fides increased dramatically toward the shore, however, solids representative of the absence of sulfate reduction 1....



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